

LEACHING OF METALS FROM SECONDARY LEAD SMELTER MATTE

F. Kukurugya, E. Kim, P. Nielsen, L. Horckmans, J. Spooren, K. Broos, M. Quaghebeur



VITO VISION ON TECHNOLOGY



















https://vito.be/en



UNIT SUSTAINABLE MATERIALS (SUMAT)





SUSTAINABLE MATERIALS

Our expertise



Design of ceramic and metallic components



Surface treatment with plasma technology



Circular business models



Policy for the transition to a green circular economy



Material analysis and tests





TECHNOLOGICAL CORE: WASTE TO PRODUCT (WtP)

from WASTE industrial residues & wastes: inorganic sludges ashes slags construction & demolition shredder urban & landfill mines

over VALUABLES

INE 1

LINE 2

SEAR(

RE.

SEARC

separation and recovery by physico-chemical (pre-)treatment technologies including physical separation and hydrometallugy







to PRODUCTS

matrix valorisation into new products via sustainable binder and shaping technologies for residues.









Innovations in hydrometallurgy

Challenges

- Energy efficiency ↑
- Leaching efficiency ↑
- Leaching kinetics ↑
- Selectivity ↑
- Environmental friendly
- Simple process (simultaneous leaching and recovery)
- Cost effective



Methods for advanced leaching

- Microwave assisted leaching
- Microwave assisted roasting/heating/milling
- Mechanical activation
- Ultrasonic assisted leaching
- Alkali roasting



MAXIVIA PROJECT





SECONDARY LEAD PRODUCTION





SLAG AND MATTE







Slag is broken and sieved, 2/3 (~> 15 cm diameter) is reused in process.

Remainder of slag and iron stone is currently disposed = materials studied in MaxiVIA



MATERIAL CHARACTERIZATION





LEACHING OF FM FRACTION OF MATTE



Kim et al. : Hydrometallurgy 169 (2017) 290–296

Kim et al. (2017), Hydrometallurgy, In press, Accepted manuscript: <u>http://dx.doi.org/10.1016/j.hydromet.2017.02.027</u>



EFFECT OF MILLING TIME

- milling is a common pretreatment operation prior to the leaching step
- to liberate Pb phases and increase the surface area = beneficial for leaching
- dry vs. wet milling and their effect on leaching efficiencies

<u>0.5M HNO₃ + 0.15 Fe(NO₃)₃; L:S = 10 (20 ml/2g); 200 rpm; 25°C ; 90 min.</u>

- conditions chosen based on the previous studies





MILLING VS. LEACHING EFFICIENCY





PARTICLE VS. AGGLOMERATE



EPMA analysis





XRD RESULTS







CHEMICAL VS. ELECTROCHEMICAL DISSOLUTION

HOW TO AVOID AGGLOMERATION ?

Wet milling planetary ball mill; 400 rpm







citrate leaching is more selective

🧡 vito

Kukurugya et al.(2017), submitted to Hydrometallurgy

CONCLUSION

- Leaching of metals from a matte can be significantly affected by milling time
- After dry milling longer than 5 min. (at 400 rpm), agglomeration rate > breakage rate
- Pb FeS agglomeration induces a galvanic effect = 1 Fe dissolution and Pb precipitation as PbS
- To avoid agglomeration, wet milling can be used = finer particles = ↑ reaction area
- After wet milling and citrate leaching 94% of Pb can be selectively leached out
- Solid residue can be used (after treatment) as secondary iron ore



Thank you for your attention !



